STUDIES ON β -LACTAM ANTIBIOTICS

I. SYNTHESIS AND *IN VITRO* ANTI-PSEUDOMONAL ACTIVITY OF 3-ISOTHIAZOLE-CEPHALOSPORIN DERIVATIVES

NORIAKI NAGANO, KOHJI NAKANO, TADAO SHIBANUMA, Yukiyasu Murakami and Ryuichiro Hara

Central Research Laboratories, Yamanouchi Pharmaceutical Co., Ltd., 1-1-8, Azusawa, Itabashi-ku, Tokyo 174, Japan

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The synthesis and *in vitro* activity of 7β -[(Z)-2-(2-amino-4-thiazolyl)-2-(2-carboxy-2alkoxyimino)acetamido]cephalosporins with a (4-carboxy-3-hydroxy-5-isothiazolyl)thiomethyl group at the 3-position are described. These cephalosporins ($9a \sim 9i$) showed excellent activity against Gram-negative bacteria including β -lactamase producing strains. The most interesting compound of the series was 7β -[(Z)-2-(2-amino-4-thiazolyl)-2-(2-carboxy-2-propoxyimino)acetamido]-3-cephem-4-carboxylic acid (9g, YM-13115) because of its outstanding inhibitory potency against *Pseudomonas aeruginosa* and highly prolonged plasma half-life in rats.

At present, the opportunistic infectious diseases have progressively increased and become a serious problems in chemotherapy. These diseases have been mainly caused by various Gram-negative bacteria such as *Pseudomonas aeruginosa*, *Escherichia coli*, *Klebsiella pneumoniae*, *Proteus* sp. and *Serratia marcescens*, *etc.* Ceftazidime has been reported to be a very broad spectrum antibacterial agent with high degree of resistance to most β -lactamases and potentially active against *P. aeruginosa*¹⁾. It is a parenteral aminothiazolyl cephalosporin bearing a pyridinium methyl group at the 3-position. In a previous paper²⁾, we reported the synthesis and antimicrobial activity of a semi-synthetic cephamycin, cefotetan (YM-09330). As shown in Fig. 1, the tautomer of cefotetan possessing an isothiazole ring on 7β -substituent also had potent broad-spectrum antibacterial activity. This paper describes

the synthesis and *in vitro* activity of several kinds of cephems $(9a \sim 9i)$ bearing an isothiazole ring at the 3-position. These compounds had high activities against Gram-negative bacteria, especially against strains of *P. aeruginosa*, and showed long plasma half-lives in rats.

Chemistry

As one of the starting materials, 7β -amino-3-(4-carboxy-3-hydroxy-5-isothiazolyl)-3-cephem-4-carboxylic acid (3) was prepared from trisodium 4-carboxy-3-hydroxy-5-mercaptoisothiazole and a 3-iodomethylcephem compound (2) which was prepared in one-pot process from iodination of 7-ACA (1). This iodination method³⁾, which was developed in our laboratory, was carried





Scheme 1.

out with sodium iodide and trifluoromethanesulfonic acid. The other starting materials, (Z)-2-(2-tritylamino-4-thiazolyl)-2-(*tert*-butoxycarbonylalkoxyimino)acetic acids (**7b**, **7c**, **7d**, **7e**, **7i**) were prepared from 4 by alkylation with α -bromo-*tert*-butyl esters (**5b**, **5c**, **5d**, **5e**, **5i**), followed by subsequent alkaline hydrolysis as shown in Scheme 1, and the other acids (**7a**, **7f**, **7g**, **7h**) were obtained in a similar manner as reported by the Glaxo group⁴⁾ and the Roussel-Uclaf group⁵⁾. The coupling of the acids (**7a** ~ **7i**) to the 7 β -aminocephalosporin nucleus (**3**) was accomplished *via* their active esters, formed by well-known method such as DCC-HOBT. Stepwise removal of two protective groups of **8a**~**8i** with trifluoroacetic acid-anisol and trifluoroacetic acid - water gave the products (**9a**~**9i**) as shown in Scheme 2.

Biological Results

The *in vitro* antibacterial activities of the new 3-[(4-carboxy-3-hydroxy-5-isothiazolyl)thiomethyl]cephem compounds (9a~9i) against selected Gram-positive and Gram-negative organisms were shown in Table 1. The antibacterial activities of 9a~9i against *Staphylococcus aureus* were clearly inferior to those of ceftriaxone and ceftazidime. Against most of the Gram-negative bacteria, the activities were similar. However, against *P. aeruginosa*, 9a~9i were clearly more active than ceftazidime. Especially, 9g (YM-13115)^{8,7)}, 9h and 9i showed very strong activities against *P. aeruginosa* and their activities were 4 to 16 times as strong as that of ceftazidime. Furthermore, the relationship between MIC of 9a~9i against *P. aeruginosa* and substituent groups (R₁ and R₂) of 9a~9i was investigated. In Fig. 2, the mean MIC of 9a~9i against *P. aeruginosa* was plotted against the values of substituent lipophilicity $(\pi)^{8)}$ of R₁ and R₂ of 9a~9i. When R₁ is hydrogen, a significant parabolic relationship is observed



between the anti-pseudomonal activity and the values of π . The compounds which showed the strongest *in vitro* activity against *P. aeruginosa*, had π values in the range of 1.4 to 1.8. When R₁ is methyl, the increasing lipophilicity of R₂ (Me, Et and *n*-Pr) has little influence on the anti-pseudomonal activity, and all of the compounds have strong activities against *P. aeruginosa*. The value for substituent lipophilicity of **9g** was 1.78. The plasma half-lives in rats^{e)} were 48 minutes for **9g**, 34 minutes for ceftriaxone, 33 minutes for **9h** and 14 minutes for ceftazidime. The stated above higher inhibitory potency of **9g** against *P. aeruginosa* and its markedly longer plasma half-life in rats in comparison with those of ceftazidime seem to be attributed to the difference in 3-position moiety of cephem ring.

Experimental

NMR spectra were recorded at 90 MHz on a Jeol 90-Q spectrometer and at 100 MHz on a Jeol MH100NMR spectrometer using tetramethylsilane as an internal standard. IR spectra were taken on a Hitachi 260-10 spectrophotometer. For column chromatography, silica gel (Wakogel C-200) was used. Melting points of the cephalosporins are not accurately reproducible because of extensive decomposition.

<u>General Preparation of (Z)-2-(2-Tritylamino-4-thiazolyl)-2-(tert-butoxycarbonylalkoxyimino)-</u> acetic Acid (7)

To a solution of ethyl (Z)-2-hydroxyimino-2-(2-tritylamino-4-thiazolyl)acetate hydrochloride (4,

	MIC (µg/ml)											
Strain	a	b	c	d	e	f	g	h	i			
R_{1}/R_{2} :	H/H	H/Me	H/Et	H/n-Pr	H/n-Bu	H/Ph	Me/Me	Me/Et	Me/n-Pr	CAZ	CTRX	
Staphylococcus aureus Smith	50	50	50	25	25	25	50	50	50	12.5	3.13	
Escherichia coli O-1	≦0.2	≦0.2	≦0.2	≦0.2	≦0.2	≤ 0.2	≦0.2	≦0.2	≤0.2	≤0.2	≤ 0.2	
E. coli Ebara	≤ 0.2	≦0.2	≦0.2	≦0.2	≦0.2	≤ 0.2	≤0.2	≤ 0.2	≤ 0.2	0.78		
Klebsiella pneumoniae ATCC 10031	≦0.2	≦0.2	≦0.2	≦0.2	≦0.2	≦0.2	≤ 0.2	≤ 0.2	≤ 0.2	≤ 0.2	≤ 0.2	
K. pneumoniae V-17	3.13	≦0.2	≦0.2	≦0.2	≤0.2	1.56	≤ 0.2	≤0.2	≤ 0.2	0.39	0.78	
Serratia marcescens IID-620	≦0.2	1.56	≦0.2	0.39	0.78	1.56	≤ 0.2	≤ 0.2	≤ 0.2	0.39	0.78	
Proteus rettgeri Y-1	≤ 0.2	0.39	0.39	0.39	0.78	≤0.2	≤ 0.2	≤ 0.2	≤ 0.2	3,13	<0.2	
Enterobacter cloacae 963 (MS-1)		0.78	0.78	0.78	1.56	1.56	0.39	0.78	≤ 0.2	0.39	<u>≡</u> 0.2	
E. aerogenes NY-2	3.13	1.56	0.78	0.78	1.56	3.13	0.78	0.78	≤ 0.2	6 25	0.78	
Pseudomonas aeruginosa NCTC 10490	0.78	≦0.2	≦0.2	≦0.2	≤ 0.2	1.56	≤0.2	≤ 0.2	0.39	0.78	0.78	
P. aeruginosa IID 5142	1.56	0.39	≦0.2	0.39	0.39	0.78	<u>≤0.2</u>	≤ 0.2	<0.2	3 13	6.25	
P. aeruginosa NC-5	3.13	0.78	0.39	1.56	3.13	0.39	0.39	≦°.2 ≤0.2	<u>≡</u> 0.2 <0.2	3 13	50	
P. aeruginosa 99	1.56	0.78	0.39	1.56	3.13	3.13	0.78	<u>≤</u> 0.2	0.39	3.13	100	

Table 1. Antibacterial activities of $3-[(4-\operatorname{carboxy-}3-\operatorname{hydroxy-}5-\operatorname{isothiazolyl})$ thiomethyl]cephem compounds ($9a \sim 9i$).

CAZ: Ceftazidime, CTRX: ceftriaxone.

Fig. 2. Relationship between the anti-pseudomonal activity and π of R_1 and R_2 of $9a \sim 9i$.



Table 2.	Yields.	mp.	analytical	and IR	data	of 7b,	7c,	7d,	7e a:	nd	7i
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Compounds		s	Yield (from 4)	Anal Calcd for Calcd:	IR (KBr) cm ⁻¹	
No.	No. R_1 R_2		mp (°C)	Found:	COO [*] Bu	
				$C_{31}H_{31}N_3O_5S$ (557.63)		
7b	н	Me	20% 166~167	C 66. 77, H 5. 54, N 7. 54, S 5. 75 C 66. 68, H 5. 85, N 7. 29, S 5. 74	1710	
			44%	$C_{32}H_{33}N_{3}O_{5}S \cdot \frac{1}{4}H_{2}O (576.20)$ C 66.71, H 5.85, N 7.29, S 5.56		
7c	Η	Et	172~173	C 66.84, H 5.81, N 7.31, S 5.63 $C_{33}H_{35}N_{3}O_{5}S$ (587.72)	1735	
		_	43%	C 67.67, H 6.02, N 7.17, S 5.47	1725	
7d	Η	<i>n</i> -Pr	168~169	C 67.63, H 6.06, N 7.22, S 5.51 $C_{34}H_{37}N_{3}O_{5}S$ (599.75)	1735	
			31%	C 68.09, H 6.22, N 7.01, S 5.35	1725	
7e	H	<i>n</i> -Bu	102~105	C 68.25, H 6.00, N 6.98, S 5.39 $C_{34}H_{37}N_{3}O_{5}S$ (599.75)	1735	
			30%	C 68.09, H 6.22, N 7.01, S 5.35		
7i	Me	<i>n</i> -Pr	175~176	C 67.83, H 6.43, N 6.72, S 5.37	1735	

Table 3. ¹H NMR spectral data of 7b, 7c, 7d and 7e.

Com	pounds		¹ H NMR δ v			
No.	\mathbf{R}_2	NH	Thiazole 5-H	H C R ₂	<i>tert-</i> Bu	R_2
7h	Me	8.76	6.80	4.52	1.38	1.33 (3H, d)
7c	Et	8.82	6.84	4.37	1.40	0.88 (3H, t), 1.72 (2H, m)
7d	<i>n</i> -Pr	8.81	6.84	4.43	1.40	0.90 (3H, t), $1.22 \sim 1.51 (2H, m),$ $1.54 \sim 1.80 (2H, m)$
7e	<i>n-</i> Bu	8.80	6.83	4.21	1.35	0.88 (3H, t), 1.12~1.55 (4H, m), 1.55~1.83 (2H, m)

0.005 mol) in DMSO (20 ml) was added powdered potassium carbonate (1.0 g) and *tert*-butyl α bromoalkylacetate (5, 0.006 mol). After being stirred at room temp for one night, the mixture was partitioned between EtOAc and ice-water. The organic layer was washed with brine, dried (MgSO₄)

	Compounds				¹ H N	$MR \delta$ value (I	OMSO- <i>d</i> ₆) ppm		
No.	R ₁	R ₂	CONH (1H, d)	Thiazole 5-H (1H, s)	C(7)-H (1H, dd)	C(6)-H (1H, d)	C(3)-CH ₂ (2H, q)	<i>tert</i> -Bu (9H, s)	R_1, R_2
8a	Н	Н	9.38	6.72	5.69	5.13	4.16	1.42	4.50 (2H, s)
8b	н	Me	9.32,	6.71,	5.72	5.15	4.18	1.39	4.49 (1H, q),
			9.41	6.72					1.27 (3H, d)
8c	н	Et	9.40	6.74,	5.74	5.21	4.22	1.44	4.38 (1H, t),
				6.78					1.70, 0.88
8d	Н	<i>n</i> -Pr	9.44	6.72,	5.74	5.18	4.22	1.40	4.39 (1H, t),
				6.74					1.2~1.9, 0.88
8e	н	<i>n</i> -Bu	9.43	6.73,	5.74	5.18	4.20	1.40	4.39 (1H, t),
				6.71					1.2~1.9, 0.88
8f	Н	Ph	9.54	6.76,	5.69	5.11	4.15	1.36	5.44 (1H, s),
				6.73					7.38 (5H)
8g	Me	Me	9.26	6.60	5.68	5.15	4.16	1.36	1.38 (6H, s)
8h	Me	Et	9.27	6.66,	5.71	5.13	4.16	1.39	1.22 (3H, s),
				6.64					1.71, 0.84
8i	Me	<i>n</i> -Pr	9.34,	6.66,	5.72	5.20	4.20	1.40	1.40 (3H, s),
			9.40	6.68					1.0~1.9, 0.84

Table 4. ¹H NMR spectral data of 8.

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1	Compounds	npounds ¹ H NMR δ value (DMSO- d_6) ppm							
No.	R ₁	R ₂	CONH (1H, d)	Thiazole 5-H (1H, s)	C(7)-H (1H, dd)	C(6)-H (1H, d)	C(3)-CH ₂ (2H, q)	C(2)-2H (2H, q)	$\mathbf{R}_1, \mathbf{R}_2$
9a	Н	Н	9.44	6.76	5.78	5.17	4.16	3.64	4.57 (2H, s)
9b	Н	Me	9.48, 9.44	6.80, 6.79	5.93, 5.79	5.22	4.20	3.69	1.40 (3H, d)
9c	н	Et	9.48, 9.46	6.78, 6.76	5.96, 5.72	5.21	4.22	3.68	4.49 (1H), 1.81 (2H), 0.96 (3H, t)
9d	H	<i>n</i> -Pr	9.48	6.78, 6.76	5.98, 5.72	5.22	4.22	3.70	4.54 (1H), 1.98~1.2, 0.91 (3H, t)
9e	Н	<i>n</i> -Bu	9.48	6.78, 6.77	5.98, 5.72	5.22	4.20	3.68	4.50 (1H), 1.98~1.2, 0.86 (3H, t)
9f	Н	Ph	9.60, 9.50	6.82, 6.78	5.95, 5.68	5.18, 5.15	4.18	3.55	5.57 (1H), 7.64~7.10
9g	Me	Me	9.38	6.70	5.82	5.17	4.15	3.66	1.41 (6H, s)
9h	Me	Et	9.38	6.71	5.90, 5.70	5.18	4.18	3.68	1.31, 1.41, 1.82, 0.86 (3H, t)
9i	Me	<i>n</i> -Pr	9.36	6.72	5.96, 5.72	5.20	4.16	3.66	$1.04 \sim 1.92$, 0.84 (3H, s)

Table 5. ¹H NMR spectral data of 9.

and evaporated under reduced pressure. The residue was triturated with *n*-hexane to afford ethyl (Z)-2-(2-tritylaminothiazolyl)-2-(*tert*-butoxycarbonylalkoxyimino)acetate (6). A solution of above diester (6) and powdered potassium carbonate (500 mg) in methanol (55 ml) and water (6 ml) was refluxed for $4 \sim 5$ hours. The mixture was concentrated to about 6 ml and extracted with EtOAc after acidification (pH 1.0) with 1 N HCl. The separated organic layer was washed with brine, dried (MgSO₄), and then evaporated under reduced pressure. The residue was triturated with Et₂O to afford acid 7.

<u>Preparation of 7- β -Amino-3-[(4-carboxy-3-hydroxy-5-isothiazolyl)thiomethyl]-3-cephem-4-carboxylic Acid (4)</u>

To a solution of the 7-amino-3-iodomethylcephem compound (34 g, 0.1 mol) and NaHCO₃ (8.4 g, 0.1 mol) in water (600 ml) was added trisodium 4-carboxy-3-hydroxy-5-mercaptoisothiazole (26.7 g, 1.11 mol) and the mixture was stirred for 3 hours at room temp. The reaction solution was adjusted to pH 1.6 with 2 N HCl (114 ml) under ice-cooling, the separated solid was collected by suction and washed with cold water and dried (P_2O_5) to give 3 as pale brown powder (29 g, 74.6%): MP 170~ 200°C (dec); IR (KBr) 1770 cm⁻¹; NMR ($D_2O+NaHCO_3$) δ 3.58 (2H, q, 2-CH₂), 3.98 (2H, q, 3-CH₂), 5.04 (1H, d, 6-H), 5.41 (1H, d, 7-H).

<u>General Preparation of 3-[(4-Carboxy-3-hydroxy-5-isothiazolyl)thiomethyl]-7 β -[(Z)-2-(2-trityl-amino-4-thiazolyl)-2-(*tert*-butoxycarbonylalkoxyimino)acetamido]-3-cephem-4-carboxylic Acid (8)</u>

A mixture of 7 (3.3 mmol), 1-hydroxybenzotriazole (3.6 mmol) and DCC (3.6 mmol) in dioxane (16.6 ml) was stirred for 1 hour at room temp. The mixture was filtered to remove a small amount of insoluble material and the filtrate was added to a solution of 3 (3 mmol) and NaHCO₃ (6.2 mmol) in water (10 ml). After stirring at room temp for one night, the mixture was concentrated under reduced pressure. The residue was treated with 5% aqueous NaHCO₃ (10 ml) and washed with EtOAc. The aqueous layer was then extracted with methyl ethyl ketone (MEK) after acidification (pH 1.5) with 2 N HCl. The separated organic layer was washed with brine, dried (MgSO₄) and then evaporated under reduced pressure without heating. The residue was fractionated by silica gel chromatography (CHCl₃ - 2-PrOH - HCOOH, 90: 10: 2) to give, after trituration with mixture of Et₂O and *n*-hexane, **8** as a powder.

<u>General Preparation of 3-[(4-Carboxy-3-hydroxy-5-isothiazolyl)thiomethyl]-7 β -[(Z)-2-(2-amino-4-thiazolyl)-2-(hydroxycarbonylalkoxyimino)acetamido]-3-cephem-4-carboxylic Acid (9)</u>

Trifluoroacetic acid (6 ml) was added to a mixture of 8 (0.589 mmol) in anisol (0.5 ml) under ice-cooling, and then mixture was stirred for 1 hour at $15 \sim 20^{\circ}$ C. After removing of the trifluoroacetic acid under reduced pressure without heating, the residue was triturated with Et₂O. The collected precipitate was added in a mixture of trifluoroacetic acid (6 ml) and water (2 ml) under ice-cooling. After being stirred at $15 \sim 20^{\circ}$ C for 1 hour, the trifluoroacetic acid and water were evaporated under reduced pressure. The residue was triturated with Et₂O to give 9 as a powder.

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